after the salt and sample are fused in, then taking the difference. Particularly with old inserts there is some extrusion and  $t_{\rm C}$  at the end of the run is slightly different than the initial  $t_{\rm C}$ . (It is essential to micrometer both the assembly and individual pistons at the end as there is measurable shortening of the pistons). Where to varied appreciably from the initial to the final value, it was assumed that the effective  $t_{\rm C}$  varied linearly with  $p_{\rm A}$ , reaching its final value at the highest  $p_{\rm A}$ . Equations (2) and (3) have been tested for values of  $t_{\rm C}$ ranging from 4-15 (0.004" - 0.015").

The pressures are necessarily less accurate in Cell II, so it is desirable to make all studies in both cells allowing overlap for continuous calibration. It is difficult to estimate exactly the accuracy of the pressure determination. The transitions of the silver halides were reproducible to  $\pm$  1000 atmospheres. Equations (2) and (3) reproduced a large number of experimental points with the largest deviation in t<sub>0</sub> being 0.1 (0.0001") as long as  $p_c$  was greater than 50 (50,000 atmospheres).

Figure 3 shows the press constructed for this work. The dimensions can, of course, be varied, but we used a body of AISI 6150 hardened to 45-46 Rockwell C 6" O.D. and 4 1/2" I.D. The windows are 2" X 2 1/2". It is necessary to back the pistons with carboloy blocks 1" in diameter and one inch thick. The press is portable and can be inserted at the normal sample point of many spectrometers with minor modifications of the optics.

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